

## Receptor-mediated Targeting of Fluorescent Probes in Living Cells\*

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A strategy was developed to label specified sites in living cells with a wide selection of fluorescent or other probes and applied to study pH regulation in Golgi. cDNA transfection was used to target a single-chain antibody to a specified site such as an organelle lumen. The targeted antibody functioned as a high affinity receptor to trap cell-permeable hapten-fluorophore conjugates. Synthesized conjugates of a hapten (4-ethoxymethylene-2-phenyl-2-oxazolin-5-one, phOx) and fluorescent probes (Bodipy FL, tetramethylrhodamine, fluorescein) were bound with high affinity (~5 nM) and specific localization to the single-chain antibody expressed in the endoplasmic reticulum, Golgi, and plasma membrane of living Chinese hamster ovary cells. Using the pH-sensitive phOx-fluorescein conjugate and ratio imaging microscopy, pH was measured in the lumen of Golgi (pH 6.25 ± 0.06). Measurements of pH-dependent vacuolar H<sup>+</sup>/ATPase pump activity and H<sup>+</sup> leak in Golgi provided direct evidence that resting Golgi pH is determined by balanced leak-pump kinetics rather than the inability of the H<sup>+</sup>/ATPase to pump against an electrochemical gradient. Like expression of the green fluorescent protein, the receptor-mediated fluorophore targeting approach permits specific intracellular fluorescence labeling. A significant advantage of the new approach is the ability to target chemical probes with custom-designed spectral and indicator properties.

Small-molecule fluorescent probes have been widely used to study protein localization, cytoplasmic ionic content (1), and solute diffusion (2). Probes are available with high intrinsic brightness and excitation and emission peaks from ultraviolet to infrared wavelengths; however, in general these chemical probes cannot be targeted to specific sites in living cells. The green fluorescent protein (GFP)<sup>1</sup> is a genetically targetable

probe that has been used extensively to study gene expression and protein localization in living cells (3, 4). However, GFP fluorescence is limited to blue, cyan, green, and green/yellow variants, which have relatively low intrinsic brightness (4, 5). GFP-based indicators are currently limited to measuring pH (6–8), Ca<sup>2+</sup> (9, 10), and membrane potential (11).

A cell labeling method is reported here that combines the site specificity conferred by genetically encoded targeting sequences with the excellent spectral and indicator properties of small chemical fluorophores. The strategy is to express a high affinity “receptor” at a specified intracellular location to trap a conjugate of a fluorophore linked to a receptor “ligand” (Fig. 1a). We chose a single-chain antibody (sFv) (12) as the receptor and a hapten (phOx) as the ligand. Although many receptor-ligand pairs are possible, the antibody-hapten pair was selected because of the simple ligand-probe chemistry and high affinity interaction without interference from cellular factors. For sFv targeting, cells are transfected with cDNAs encoding sFv in fusion with targeting sequences. Fluorophore-hapten conjugates are added to the extracellular solution at low concentrations, diffuse to sites of sFv expression, and bind to the sFv. Conjugates of different indicator and spectral properties were synthesized (Fig. 1b), including phOx-Bodipy FL (green fluorescent), phOx-fluorescein (green fluorescent, pH-sensitive), and phOx-tetramethylrhodamine (red fluorescent). The flexible linkers were designed to permit stacking of the unbound hapten with its covalently attached fluorophore to form a dark complex and reduce background fluorescence.

The ability to label cellular sites with fluorescent probes with varied spectral and indicator properties was demonstrated, and the phOx-fluorescein conjugate was applied to measure pH in the Golgi lumen. Here we report the first *in vivo* measurement of the regulation of the pump rate of the vacuolar H<sup>+</sup>/ATPase by Golgi luminal pH to test the thermodynamic model (13) of how the resting pH is set in organelles such as the Golgi.

### EXPERIMENTAL PROCEDURES

**Plasmids and Cell Transfection**—The cDNA encoding the sFv (25-kDa protein) and two c-Myc epitopes was amplified by polymerase chain reaction using the plasmid pHook1 (12) (Invitrogen) as template, with sense primer (5'-GGAATTCGCCGAGGTCAAGCTGCAGGAG3') containing an *Eco*RI site (underlined) and antisense primer containing an *Xba*I site (underlined) (5'-GCTCTAGACTGGCCACAGCATTGATCCTC3'). For ER and Golgi targeting, the cDNA was subcloned at *Eco*RI and *Xba*I sites in expression plasmid pCDNA3.1 containing specific targeting sequences as described in Ref. 6. Plasma membrane targeting was achieved with the pHook1 plasmid (12). CHO cells (ATCC CRL 9618) were transfected with plasmids encoding targeted sFv using LipofectAMINE (Life Technologies, Inc.) as described previously (6).

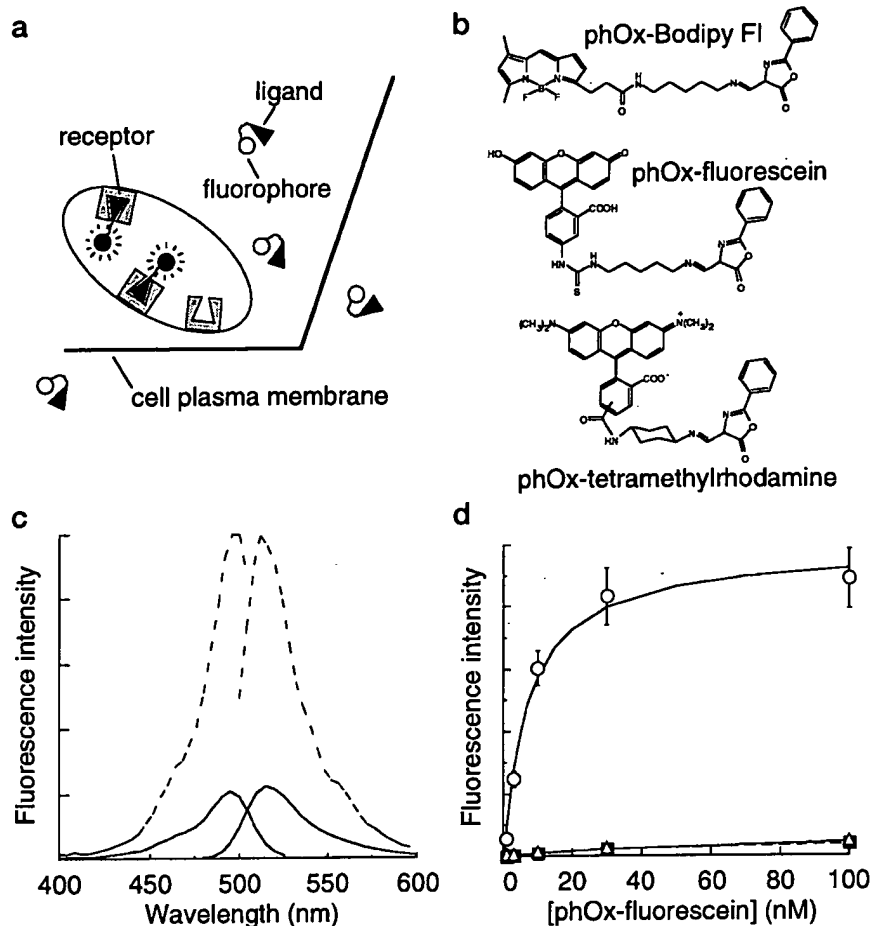
**Synthesis of Hapten-Fluorophore Conjugates**—A flexible linker was added to phOx (Sigma) by reaction of 50 mg of phOx with 14.3 μl of 1,5-diaminopentane (Aldrich) in 2.5 ml of acetone for 1 h. The disubstituted aminopentane was precipitated by the addition of 2 volumes of 50 mM borate buffer (pH 9.2), leaving the product in solution. A more rigid linker was added to phOx by reaction of *trans*-1,4-diaminocyclohexane (Aldrich) (5.1 mg in 0.5 ml of Me<sub>2</sub>SO) and phOx (10 mg in 0.5 ml of acetone) for 2 h. The disubstituted diaminocyclohexane was precipitated by the addition of an equal volume of water, leaving the product in solution. phOx-Bodipy FL was prepared by reaction of excess Bodipy FL succinimidyl ester (Molecular Probes) with 1 mM phOx-aminopentane in borate buffer for 2 h. The product was obtained as a precipitate. PhOx-tetramethylrhodamine was prepared by reaction of excess tetramethylrhodamine succinimidyl ester (Molecular Probes) with 5 mM phOx-aminocyclohexane in borate buffer for 6 h. The product was obtained as a precipitate. phOx-fluorescein was prepared by reaction of

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<sup>1</sup> The abbreviations used are: GFP, green fluorescent protein; phOx, 4-ethoxymethylene-2-phenyl-2-oxazolin-5-one; ER, endoplasmic reticulum; CHO, Chinese hamster ovary.

**FIG. 1. Strategy for fluorophore targeting.** *a*, receptor is targeted to an intracellular site by cDNA transfection. Ligand-fluorophore conjugate added to the media is trapped by the receptor. Fluorescence increases upon binding to the receptor. *b*, chemical structures of the conjugates. *c*, excitation and emission spectra of free phOx-fluorescein (1 nM) in phosphate-buffered saline (solid line) and an identical concentration of phOx-fluorescein (dashed line) bound to sFv expressed at the plasma membrane of suspended CHO cells. *d*, phOx-fluorescein fluorescence was measured at indicated concentrations for regions without cells (filled squares), regions in non-sFv-expressing cells (triangles), and regions in cells expressing sFv at the plasma membrane (circles). The fluorescence of sFv-bound phOx-fluorescein was fitted to a single site binding model with  $K_d$  6.8 nM.



equimolar amounts of phOx (0.5 mg in 25  $\mu$ l of acetone) and fluorescein cadaverine (1 mg in 50  $\mu$ l of dimethylformamide; Molecular Probes) for 1 h. Phosphate-buffered saline was added, unreacted phOx was removed by hexane extraction, and the product was extracted with butanol. phOx-ethanolamine was prepared by reaction of 1 mg of phOx with 0.3  $\mu$ l of ethanolamine in 10 ml of ethanol for 1 h. Reactions were conducted at room temperature. Products were judged to be >95% pure by TLC, and structures were confirmed by mass spectrometry.

**Fluorescence Measurements**—Cells were labeled at 2 days after transfection by incubation with low concentrations (<100 nM) of the conjugates. Unless otherwise indicated, cells were observed in the absence of the conjugate in the bathing media. Images were recorded at room temperature on a K2 BIO microscope (Technical Instruments) equipped with a 60 $\times$  PlanApo objective (Nikon, N.A. 1.4), coaxial-confocal attachment, and cooled CCD camera. Dual excitation ratio images of fluorescein were acquired using 440- and 490-nm excitation filters and a 520-nm long pass emission filter. Continuous recordings of the fluorescence time course were obtained on a Nikon Diaphot epifluorescence microscope equipped with a 100 $\times$  PlanApo objective (Nikon, N.A. 1.4), a photomultiplier using a 530-nm bandpass emission filter, and an optical filter changer (model 10-C, Sutter Instrument Co.) containing 440- and 490-nm excitation filters. Cuvette fluorescence measurements were conducted on an SLM 8000c fluorometer (SLM Aminco). Spectra were recorded with 4-nm slit widths; time courses were acquired with 495-nm excitation and 510-nm long pass emission filters.

## RESULTS

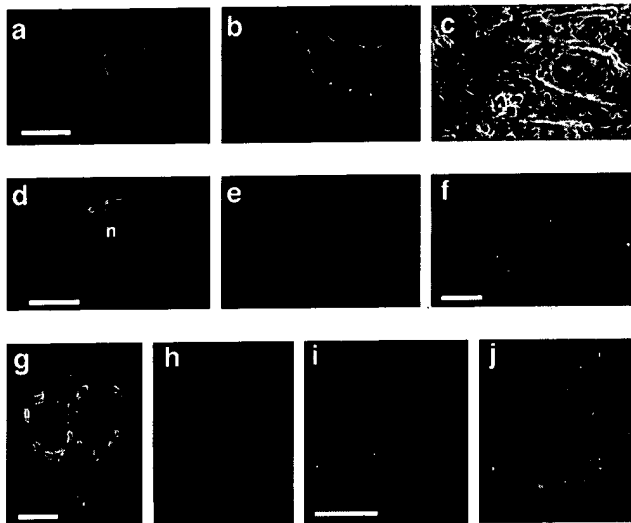
For effective targeting, the requirements of the sFv/hapten-fluorophore system include: bright fluorescence of the bound conjugate, high affinity binding of the conjugate to sFv, stability of the conjugate, minimal cellular toxicity, strong cellular expression of functional sFv, and membrane permeability of the conjugate. These requirements were fulfilled for cellular sFv expression (using the Golgi, ER, and plasma membrane vectors) and binding of the conjugates in Fig. 1c. Fluorescence

spectra of phOx-fluorescein bound to sFv and of unbound phOx-fluorescein in solution had similar spectral shapes (Fig. 1c). As intended, the fluorescence of the unbound conjugate was decreased considerably (by 5-fold) over that of the bound conjugate. Images of CHO cells expressing sFv at the plasma membrane were acquired with increasing concentration of phOx-fluorescein. The fluorescence from sFv-bound phOx-fluorescein gave a dissociation constant ( $K_d$ ) of 6.8 nM (Fig. 1d). This agrees with the value of 5.5 nM obtained in CHO cell suspensions expressing sFv at the plasma membrane (not shown). At 10 nM phOx-fluorescein, fluorescence from free dye was 75 times lower than that of bound dye. No significant fluorescence from non-sFv-expressing cells was seen.

The toxicity and stability of the conjugates were investigated. There were no differences in cell growth as assessed by cell counting and viability as assessed by trypan blue exclusion between control cells and cells incubated for 24 h with 200 nM of each conjugate. The stability of the imine bond in the conjugates was determined in cells. In freshly prepared phOx-fluorescein and phOx-fluorescein incubated with cell suspensions for up to 7 h, the imine bond was hydrolyzed in 0.5 M NaOH, resulting in increased fluorescence. The fluorescence increase after treatment with NaOH was the same in both samples, indicating that the imine bond was not hydrolyzed in cells.

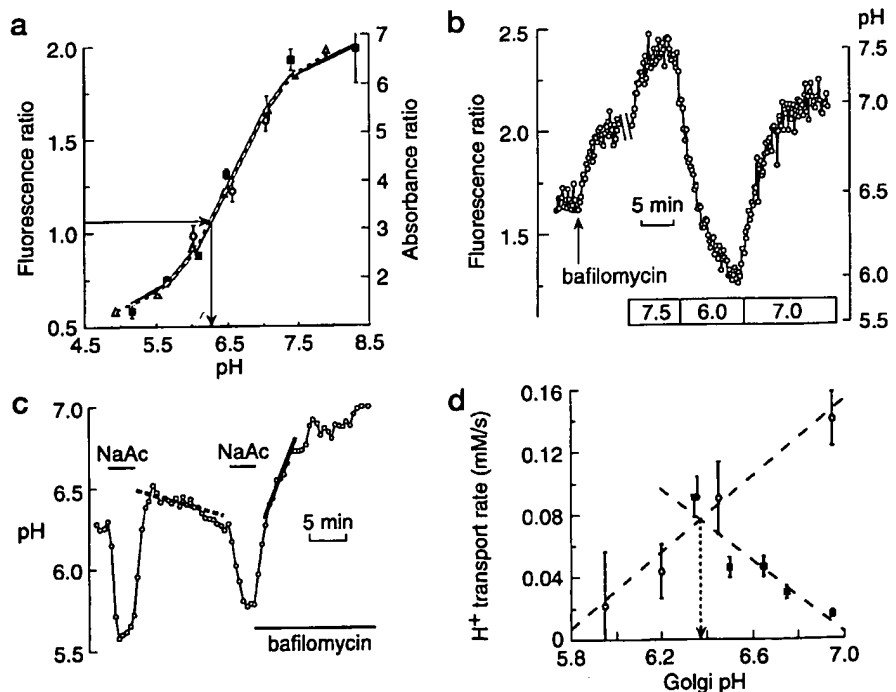
The various sFv targeting constructs and phOx conjugates were studied in CHO cells. Fig. 2a shows a fluorescence image of living cells expressing the sFv at the plasma membrane and labeled with phOx-rhodamine. A plasma membrane staining pattern was found. Fig. 2b shows staining of sFv in the same cells with a fluorescein-labeled anti-c-Myc antibody. Comparison with Fig. 2a demonstrates that only sites of sFv expression

were significantly labeled with pHx-rhodamine. There was no significant staining of adjacent cells that did not express sFv (Fig. 2c). Fig. 2d shows specific pHx-Bodipy staining of Golgi. Staining was reversed by addition of 1  $\mu$ M pHx-ethanolamine (Fig. 2e) but not by 1  $\mu$ M ethanolamine. Fig. 2f shows pHx-Bodipy staining of ER, seen as a characteristic reticular pattern. The high expression level of the sFv, the relatively high affinity of the hapten/sFv, and the low fluorescence of the



**FIG. 2. Site-specific labeling of CHO cells.** *a*, confocal fluorescence image of CHO cells transfected with the plasma membrane-targeted sFv vector and stained with pHx-rhodamine. *b*, confocal fluorescence image of cells stained with a fluorescein-labeled anti-c-Myc antibody. *c*, bright field image of cells in *a*. *d*, fluorescence image of CHO cells with Golgi-targeted sFv in the presence of 10 nM pHx-Bodipy FL (*n*, nucleus). *e*, same cells as in *d* in the presence of 1  $\mu$ M pHx-ethanolamine and 10 nM pHx-Bodipy FL. *f*, fluorescence image of cell with ER-targeted sFv in the presence of 10 nM pHx-Bodipy FL. Immunostaining of Golgi-sFv-transfected cells with fluorescein-labeled, anti-c-Myc antibody (*g*) and rhodamine-labeled anti-mouse antibody directed against a mouse 58-kDa protein antibody (*h*) is shown. Immunostaining of ER-sFv-transfected cells with rhodamine-labeled anti-mouse antibody directed against an anti-c-Myc antibody (*i*) and fluorescein-labeled concanavalin A (*j*) is shown. Scale bars, 10  $\mu$ m.

**FIG. 3. Measurement of Golgi pH.** *a*, average ratio generated by pixel-by-pixel division of 490-nm images by 440-nm images after background subtraction was converted to pH using a calibration relating pH to fluorescence signal ratios for Golgi (circles) and plasma membrane (squares) targeted pHx-fluorescein. Plasma membrane data were acquired in the presence of 10 nM pHx-fluorescein. The ratios of absorbance at 490-to-440 nm are shown for unbound pHx-fluorescein (triangles) in solution. Calibration data were fitted to a single site titration model (solid line, fluorescence; dashed line, absorbance). *b*, fluorescence ratio (490/440) time course in Golgi. At indicated times, bafilomycin A<sub>1</sub> (100 nM) and high K<sup>+</sup> buffers (120 mM) containing 5  $\mu$ M monensin at indicated pH were added. Calibration solutions were used to convert ratios to pH values (scale at right). *c*, calculated pH time course in response to a 20 mM sodium acetate prepulse in the absence and presence of 100 nM bafilomycin A<sub>1</sub>. Lines (fitted from 120 to 300 s after prepulse) indicate the initial rate of pH change. *d*, calculated proton pump rates (circles) and leak rates (filled squares) as a function of Golgi pH are shown. Dashed lines are linear fits to the data.



unbound conjugate allowed images to be obtained in the presence of <10 nM of unbound conjugate with little contribution from free conjugate. Leakage out of the Golgi, which required dissociation from the sFv and diffusion through lipid membranes and unstirred layers, had a half-time of tens of minutes.

The subcellular location of expressed sFv was confirmed by immunofluorescence. Cells transfected with the Golgi-sFv construct showed perinuclear staining by a fluorescein-labeled, anti-c-Myc antibody (Fig. 2g), which colocalized with staining by antibodies against the Golgi marker 58-kDa protein (14) (Fig. 2h). Cells transfected with the ER-sFv construct showed a reticular staining pattern with the c-Myc antibody (Fig. 2i), which colocalized with staining by fluorescein-labeled concanavalin A, an ER marker (15) (Fig. 2j). The membrane permeability of the conjugates was high enough to load cells by incubation at 37 °C for 4 h for pHx-fluorescein, 2 h for pHx-rhodamine, or 10 min for the less polar pHx-Bodipy. Cells could be loaded at 4 °C, indicating that the conjugate entered the cells primarily by transmembrane diffusion and not by endocytosis. These results demonstrate the selective targeting of fluorescent probes to expressed sFv in living cells.

Organelle-specific sFv targeting was applied to measure Golgi pH using pHx-fluorescein as the probe. Ratio images were calculated from images of Golgi labeled with pHx-fluorescein acquired at 440- and 490-nm excitation wavelengths. To convert ratios to absolute pH, cells were perfused with "calibration buffers" at different pH values containing high K<sup>+</sup> and the ionophore monensin to equalize extracellular and Golgi luminal pH. The dependence of the fluorescence ratio on pH was measured for Golgi and plasma membrane-expressed sFv (Fig. 3a). The apparent  $pK_a$  of bound pHx-fluorescein was not different from that of unbound pHx-fluorescein in solution ( $pK_a = 6.54$ ). The average Golgi fluorescence ratio of  $1.05 \pm 0.05$  corresponds to a pH of  $6.25 \pm 0.06$ , in agreement with previous estimates (7, 16, 17).

The Golgi-targeted pHx-fluorescein was used to detect continuous changes in luminal pH in individual cells. Fig. 3b shows that the fluorescence ratio increases upon addition of the vacuolar H<sup>+</sup> pump inhibitor bafilomycin A<sub>1</sub>. The ratios measured using calibration buffers were used to convert fluores-

cence ratios to pH (scale at right). Golgi pH initially at ~6.3 promptly alkalinized after the addition of bafilomycin A<sub>1</sub>.

It has been proposed that the steady-state Golgi pH is determined thermodynamically by the free energy of ATP hydrolysis used by the vacuolar H<sup>+</sup>/ATPase to pump H<sup>+</sup> against an electrochemical gradient (13). To test the model prediction that the H<sup>+</sup> pump rate is 0 at steady-state pH, the rates of Golgi H<sup>+</sup> pump and leak were measured as a function of Golgi luminal pH. After Golgi alkalinization by a 20 mM sodium acetate prepulse, H<sup>+</sup> pumping into the Golgi restores steady-state pH (Fig. 3c). The initial rate of pH change (dashed line) is the difference between pump and leak rates:  $dpH/dt = (dH^+/dt_{\text{leak}} - dH^+/dt_{\text{pump}})/\beta$ . The buffer capacity,  $\beta$ , was measured by the NH<sub>4</sub>Cl pulse method (18) to be constant ( $38 \pm 3$  mm/pH units) in the pH range 6–7 (not shown). The H<sup>+</sup> leak rate,  $dH^+/dt_{\text{leak}}$ , was measured from the pH change (Fig. 3c, solid line) after an identical 20 mM sodium acetate prepulse with the H<sup>+</sup> pump inhibited by bafilomycin A<sub>1</sub>. Similar prepulse measurements were done at different Golgi pH by varying sodium acetate and NH<sub>4</sub>Cl concentrations. There should be little effect of acetate or NH<sub>4</sub><sup>+</sup> on H<sup>+</sup> transport in the prepulse method because these ions have left the cell by the time of the H<sup>+</sup> transport measurements (18). Fig. 3d shows that the computed H<sup>+</sup> pump rate increases sharply with Golgi pH and that the pump rate is not 0 at the Golgi steady-state pH of ~6.25. Although it is recognized that the prepulse method used to alter Golgi pH also alters cytoplasmic pH, it has been reported that the H<sup>+</sup> pump rate is relatively insensitive to cytoplasmic pH in both mammalian (17) and plant (19) systems.

In the steady state, H<sup>+</sup> pump rate must equal H<sup>+</sup> leak rate. The dependence of the leak rate on Golgi pH was measured from the kinetics of pH change after bafilomycin A<sub>1</sub> addition as shown in Fig. 3b. The data for different pH are summarized in Fig. 3d, showing decreased H<sup>+</sup> leak as Golgi pH increases. The intersection of the H<sup>+</sup> pump and leak curves predicts correctly the observed steady-state Golgi pH, supporting a balanced pump/leak mechanism for setting Golgi pH.

#### DISCUSSION

The fluorophore targeting method reported here provides a new strategy for cellular labeling that complements existing methods. Of the current methods, GFP is the easiest to use because it does not require exogenously added reagents or cofactors. However, the available GFP mutants are limited in terms of spectral properties (brightness and excitation and emission wavelengths) and indicator sensitivities. Recently, a method was developed for covalently labeling proteins in living cells with a fluorescein derivative (FLASH) (20). Addition of an arseno-fluorescein derivative to the cells leads to covalent attachment of the derivative to a short  $\alpha$ -helix containing 4 cysteines added to the protein of interest. An advantage of FLASH over GFP and the current approach is the significantly smaller size of the protein tag; however, FLASH is currently limited to the use of fluorescein as the probe.

The addition of labeled macromolecules either requires microinjection (21) or is limited to compartments that are accessible by endocytosis (22) or retrograde transport through the secretory pathway (17). GFP, FLASH, and the receptor-mediated targeting methods use genetically encoded targeting sequences to localize fluorophores to virtually any cellular site, provided that the targeted protein is able to fold properly. Although strong sFv expression was found in plasma membranes and various intracellular compartments (Fig. 2, above), preliminary experiments suggest that functional sFv expression is relatively poor in reducing environments (data not shown). If disulfide bond formation is critical to sFv folding, it may be possible to generate sFv mutants lacking disulfide

bonds that fold well in reducing environments (23).

The ability to target fluorophores of varied spectral properties is a distinct advantage of the receptor-mediated targeting approach over GFP. The receptor-mediated targeting method utilizes fluorescent probes with a potentially wide range of excitation and emission wavelengths and other optical properties. The use of multiple probes with well separated excitation and emission spectra allows simultaneous labeling of multiple sites. For measurements of important cellular parameters, small chemical probes sensitive to ions (Ca<sup>2+</sup>, Na<sup>+</sup>, K<sup>+</sup>, H<sup>+</sup>, Cl<sup>−</sup>), viscosity, and membrane potential are available.

Receptor-mediated targeting of a hapten-fluorescein conjugate was used to label the Golgi with a fluorescent pH indicator permitting the measurement of the dependence of the vacuolar H<sup>+</sup>/ATPase pump rate on Golgi pH. The pump rate increased and the leak rate decreased as Golgi pH increased. The steep dependence of pump rate on pH is in agreement with measurements of ATPase activity (24) of the vacuolar H<sup>+</sup>/ATPase made *in vitro* and of the pump rate made in phagosomes (25). In contrast to the predictions of a thermodynamic model of pH regulation in the Golgi, the net pump rate was not 0 at the resting Golgi pH. Thus, the resting pH is determined by the kinetics of proton leak *versus* pump. Shifts in the leak or pump curves could account for the differences in resting pH in organelles of the secretory pathway.

In summary, the receptor-mediated probe targeting strategy allows the labeling of specified cellular structures with fluorescent or other indicator molecules. This targeting method can readily be extended to deliver conjugates containing magnetic resonance probes, caged compounds, or chemical cross-linkers. Finally, the use of cell-specific promoters and gene transfer should allow the *in vivo* targeting of hapten-probe complexes to specific cell types in multicellular organisms.

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## Green Fluorescent Protein as a Noninvasive Intracellular pH Indicator

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**ABSTRACT** It was found that the absorbance and fluorescence of green fluorescent protein (GFP) mutants are strongly pH dependent in aqueous solutions and intracellular compartments in living cells. pH titrations of purified recombinant GFP mutants indicated >10-fold reversible changes in absorbance and fluorescence with  $pK_a$  values of 6.0 (GFP-F64L/S65T), 5.9 (S65T), 6.1 (Y66H), and 4.8 (T203I) with apparent Hill coefficients of 0.7 for Y66H and ~1 for the other proteins. For GFP-S65T in aqueous solution in the pH range 5–8, the fluorescence spectral shape, lifetime (2.8 ns), and circular dichroic spectra were pH independent, and fluorescence responded reversibly to a pH change in <1 ms. At lower pH, the fluorescence response was slowed and not completely reversed. These findings suggest that GFP pH sensitivity involves simple protonation events at a pH of >5, but both protonation and conformational changes at lower pH. To evaluate GFP as an intracellular pH indicator, CHO and LLC-PK1 cells were transfected with cDNAs that targeted GFP-F64L/S65T to cytoplasm, mitochondria, Golgi, and endoplasmic reticulum. Calibration procedures were developed to determine the pH dependence of intracellular GFP fluorescence utilizing ionophore combinations (nigericin and CCCP) or digitonin. The pH sensitivity of GFP-F64L/S65T in cytoplasm and organelles was similar to that of purified GFP-F64L/S65T in saline.  $NH_4Cl$  pulse experiments indicated that intracellular GFP fluorescence responds very rapidly to a pH change. Applications of intracellular GFP were demonstrated, including cytoplasmic and organellar pH measurement, pH regulation, and response of mitochondrial pH to protonophores. The results establish the application of GFP as a targetable, noninvasive indicator of intracellular pH.

### INTRODUCTION

The green fluorescent protein (GFP) from the jellyfish *Aequorea victoria* is used widely as a noninvasive fluorescent marker for gene expression, protein localization, and intracellular protein targeting (Gerdes and Kaether, 1996; Chalfie et al., 1994; Cubitt et al., 1995). Structural analysis by x-ray crystallography indicated that the GFP protein consists of a  $\beta$ -barrel in which the oxidized triamino acid chromophore is buried in the protein interior (Yang et al., 1996; Ormö et al., 1996). Various GFP mutants have been generated that have altered physico-chemical properties, including fluorescence excitation and emission maxima, molar absorbance, and chromophore oxidation kinetics (Heim et al., 1994, 1995; Andersen et al., 1996; Cormack et al., 1996; Kimata et al., 1997). GFP has been expressed in bacteria, plants, yeast, mammalian cells, and whole organisms (Lim et al., 1995; Zolotukhin et al., 1996; Hampton et al., 1996). GFP has also been expressed in subcellular organelles by fusion with appropriate targeting sequences (Rizzuto et al., 1995, 1996; DeGiorgi et al., 1996; Girotti and Banting, 1996; Terasaki et al., 1996; Hampton et al., 1996; Cole et al., 1996; Liu et al., 1997).

GFP and selected mutants have considerable potential as noninvasive sensors of biologically important intracellular functions. Our lab used GFP as a reporter molecule to probe

the microviscosity of cytoplasm (Swaminathan et al., 1997) and mitochondria (Partikian et al., 1998) by fluorescence photobleaching and time-resolved fluorescence methods. Energy transfer between GFP mutants was exploited in the design of a protease sensor in which blue and green fluorescent GFP mutants were linked by a spacer containing a trypsin cleavage site (Heim and Tsien, 1996). A GFP-based fluorescence sensor of calcium was developed recently based on energy transfer between blue and green fluorescent GFP mutants linked by a calmodulin-binding sequence (Rommoser et al., 1997; Miyawaki et al., 1997).

The purpose of this study was to evaluate the suitability of GFP as an intracellular pH indicator. Although early studies on native GFP noted the pH sensitivity of absorbance and fluorescence (Bokman and Ward, 1981; Ward et al., 1980, 1982), the mechanism of GFP pH sensitivity has not been investigated, nor has the possibility been explored that GFP might be used as a pH sensor in living cells. We found that the fluorescence of GFP is very sensitive to pH in vitro and in vivo, and that its pH sensitivity could be modified by point mutations. GFP fluorescence responded rapidly and reversibly to pH changes. Cell experiments indicated that GFP is suitable as a noninvasive pH indicator for study of pH regulation in intracellular compartments that cannot be labeled with conventional pH indicators.

### MATERIALS AND METHODS

#### Cell culture

CHO-K1 cells (ATCC CRL9618) and LLC-PK1 cells (CL101.1) were cultured on 18-mm-diameter round glass coverslips at 37°C in 95% air, 5%  $CO_2$ . CHO cells were cultured in Ham's F12 medium, and LLC-PK1 cells in DME-H21, each containing 10% fetal bovine serum, penicillin (100

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U/ml) and streptomycin (100 µg/ml). Cells were transfected 1 day after plating (at ~80% confluence) with 1 µg of plasmid DNA encoding the various GFP fusion proteins and 12 µg of Lipofect-AMINE reagent (BRL, Bethesda, MD) in a 0.2-ml volume of OPTI-MEM (BRL). The transfection mixture was replaced with 1 ml of culture medium at 5 h. Cells were used 2–3 days after transfection.

### Plasmid constructions

The coding sequence of GFP-F64L/S65T was PCR amplified from plasmid pEGFP-C1 (Clontech) using primers: sense, 5'-GGAATTCGTGAGCAAGGGCGAGAGCTGTTAC-3'; antisense, 5'-GCTCTAGAT-TACTTGTACAGCTCGTCCATGCCG-3' (engineered *EcoRI* and *XbaI* sites underlined). The targeting construct for GFP-F64L/S65T expression in cytoplasm was prepared as described in Swaminathan et al. (1997) except for replacement of GFP-S65T by GFP-F64L/S65T. For mitochondrial targeting, the targeting presequence of subunit VIII of human cytochrome c oxidase (COX) (Rizzuto et al., 1995) was polymerase chain reaction (PCR) amplified using human kidney cDNA as template and the following primers: sense, 5'-GCCCCAAGCTTATCATGTCCGTCCTGACGCC-3'; antisense, 5'-CGGAATTCCTTCCCCTCCGGCGGC-AACG-3' (engineered *HindIII* and *EcoRI* restriction sites underlined). The COX8 targeting sequence was subcloned into eukaryotic expression vector pcDNA3.1 (Invitrogen Corp., San Diego, CA) at *HindIII/EcoRI* sites and ligated upstream and in frame with GFP-F64L/S65T at *EcoRI/XbaI* sites. For Golgi targeting, a cDNA fragment encoding residues -77 to -31 of the human  $\beta$ -1,4-galactosyltransferase (Masri et al., 1988) signal peptide was PCR amplified using human kidney cDNA as template and the following primers: sense, 5'-GCCCCAAGCTTAAGATGAGGCTTCGGAGCC-3'; antisense, 5'-CGGAATTCGCTCAGGTCGCGGC-CAGCCA-3' (engineered *HindIII* and *EcoRI* restriction sites underlined). The fragment was subcloned into pcDNA3.1 at *HindIII/EcoRI* sites, and the GFP fragment was ligated downstream and in frame at *EcoRI/XbaI* sites. For endoplasmic targeting, a cDNA fragment encoding residues -30 to 8 of bovine prolactin (Sasavage et al., 1982) was amplified using plasmid pSP-BPI as template and the following primers: sense, 5'-GCCCCAAGCTTACCATGGACAGCAAAGGTTTC-3'; antisense, 5'-CGGAATTCAGGCCCATTTGGGACAGACGG-3' (engineered *HindIII* and *EcoRI* sites underlined), and subcloned into pcDNA3.1 at *HindIII/EcoRI* sites. The GFP coding sequence was subcloned downstream and in frame as above, except that a SEKDEL sequence (Munro and Pelham, 1987) was introduced at the GFP carboxy terminus by PCR amplification using antisense primer 5'-GCTCTAGACTACAACCTCATCTTTTCTGAC TTGTACAGCTCGTCCATGCCG-3', where an engineered *Apal* site is underlined and the sequence encoding SEKDEL is in bold. Constructs were confirmed by sequence analysis.

### GFP expression and purification

GFPs S65T, T203I, Y66H, and F64L/S65T were purified by Ni-affinity chromatography after subcloning into plasmid pRSET, expression in *Escherichia coli* BL21(DE3), and bacterial lysis as described in Swaminathan et al. (1997). Proteins eluted from the affinity column were judged to be >98% pure by Coomassie-stained sodium dodecyl sulfate polyacrylamide gel electrophoresis. Proteins were dialyzed against 5 mM sodium phosphate (pH 7.4) and lyophilized.

### Spectroscopic measurements

Fluorescence spectra were recorded on an SLM 8000c fluorimeter (SLM Instruments, Urbana, IL). The fluorescence signal was normalized to a quantum counter in the reference channel. Excitation and emission monochromator slit widths were 4 nm. Fluorescence was corrected for background, dilution, and the inner filter effect as needed (generally, <3% correction). All measurements were performed at room temperature. Fluorescence lifetimes were measured on an SLM 48000 MHF fluorimeter by

multi-frequency phase-modulation fluorimetry. Excitation wavelength was 488 nm with fluorescein in 0.1 N NaOH (lifetime, 4.0 ns) as lifetime reference. Stopped-flow fluorescence measurements were done on a Hi-Tech SF51 instrument at 480 nm excitation and >515 nm emission wavelengths. A suspension of GFP-S65T in buffer A at specified initial pH was mixed with an equal volume of buffer A titrated to give a specified final pH after mixing. The instrument mixing and dead times were less than 1.5 ms. Absorbance spectra were measured on an HP8452 photodiode array spectrophotometer (Hewlett-Packard). CD spectra were acquired over 200–260 nm on a Jasco J-500A spectropolarimeter using a 0.5-mm-pathlength quartz cell.

### pH titration and quenching experiments

Titration of GFP fluorescence versus pH were performed by cuvette fluorimetry. Purified GFPs (3–10 µg/ml) were dissolved in buffer A containing (in mM): 120 KCl, 5 NaCl, 0.5 CaCl<sub>2</sub>, 0.5 MgSO<sub>4</sub>, 10 MES, 10 MOPS, 10 citrate, pH 8.00. Predetermined aliquots of 1 N HCl were slowly added with rapid stirring to titrate pH from 8.00 to 4.00 by 0.50 pH unit intervals. Fluorescence quenching titrations with acrylamide were performed using a 5 M acrylamide stock solution.

### Fluorescence microscopy

Cell fluorescence was measured using an inverted epifluorescence microscope (Nikon Diaphot). Coverglasses containing cultured cells were mounted in a laminar-flow perfusion chamber and viewed with an oil immersion objective (Nikon Plan-Apo 40X, N.A. 1.3). Cells were illuminated by a 100-W tungsten/halogen lamp powered by a stabilized direct current power supply (Oriol). A neutral density filter was used to attenuate the excitation light intensity, and an adjustable field diaphragm reduced illuminated field to a small group of cells (generally 10–20). Wavelength selection was accomplished using an HQ GFP filter set (excitation, 480 ± 20 nm; dichroic, 495 nm; emission, 510 ± 20 nm; Chroma Corp., Brattleboro, NJ). Emitted fluorescence was detected using a photomultiplier, amplifier, and analog-to-digital converter. Imaging was done on a Leitz epifluorescence microscope equipped with a Nipkow wheel coaxial confocal attachment (Technical Instruments, San Francisco, CA). Cells were viewed with a 60X oil immersion objective (Nikon Plan-Apo, N.A. 1.4) for detection of confocal fluorescence images by a cooled CCD camera.

### In vivo pH calibration

Coverslips containing GFP-F64L/S65T-transfected cells were mounted in the perfusion chamber and perfused initially for 1–5 min with PBS (in mM): 137 NaCl, 2.7 KCl, 0.7 CaCl<sub>2</sub>, 1.1 MgCl<sub>2</sub>, 1.5 KH<sub>2</sub>PO<sub>4</sub>, 8.1 Na<sub>2</sub>HPO<sub>4</sub>, pH 7.4, at 2–10 ml/min. The perfusate was then switched to a series of calibration solutions (buffer B) containing (in mM) 120 KCl, 20 NaCl, 0.5 CaCl<sub>2</sub>, 0.5 MgSO<sub>4</sub>, and 20 HEPES and 5–10 µM nigericin and 10–20 µM CCCP, pH 4.00–8.00 (in 0.5 pH unit intervals). In some experiments the ionophores were replaced by digitonin (0.005% w/v).

### RESULTS

It was found that the fluorescence and absorbance properties of various GFP mutant proteins were strongly pH dependent. Fig. 1 A shows fluorescence excitation and emission spectra of GFP-S65T. Although the shape of the spectra did not change significantly with pH, the intensities decreased progressively with lowered pH, decreasing to 50% of maximal intensity at a pH of ~6.0. An absorbance titration was done to determine whether the molar extinction of GFP-S65T was pH dependent (Fig. 1 B). Two absorbance max-

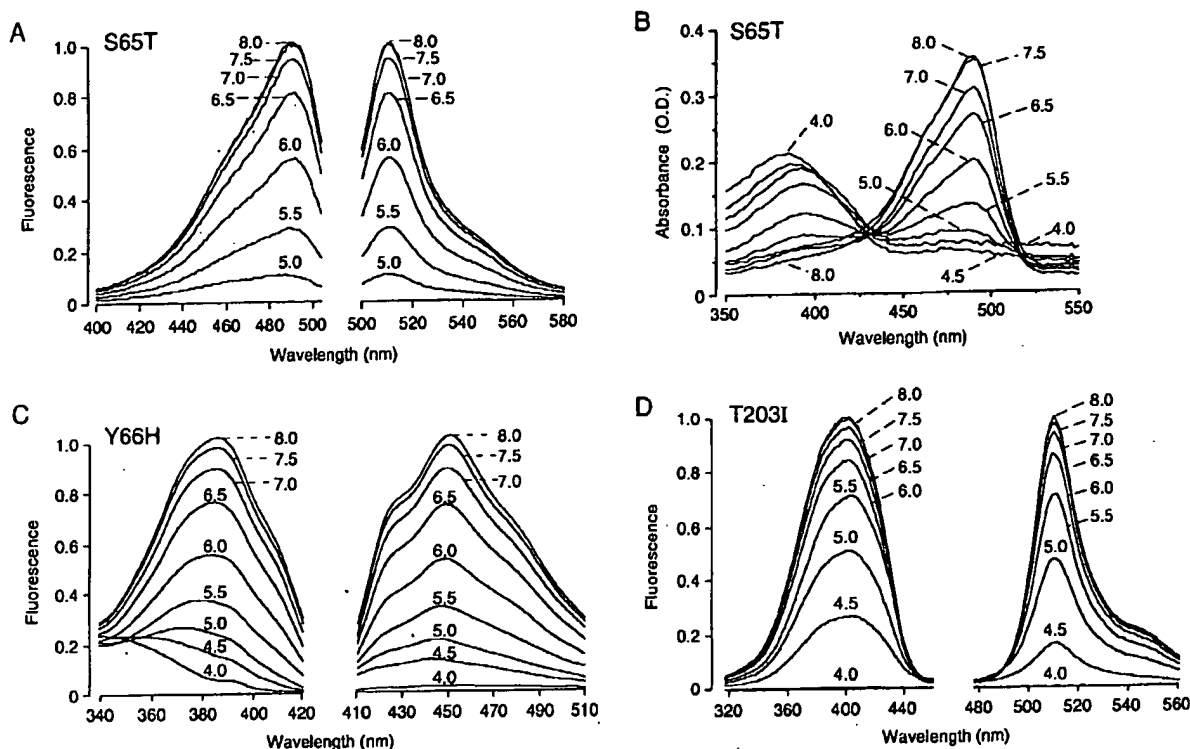


FIGURE 1 pH-dependent spectral properties of GFPs. GFPs were expressed in bacteria and purified as described in Materials and Methods. (A) Fluorescence excitation and emission spectra of GFP-S65T ( $\sim 20 \mu\text{g/ml}$ ) in buffer A at indicated pH. (B) Absorbance spectra of GFP-S65T ( $0.25 \text{ mg/ml}$ ). (C and D) Fluorescence spectra of GFP-Y66H and GFP-T203I as in A.

ima were seen. An absorbance at 490 nm, which corresponded to the fluorescence excitation maximum, decreased with pH in parallel to the fluorescence excitation. A second absorbance maximum at  $\sim 390 \text{ nm}$  increased with pH but did not excite GFP-S65T fluorescence at  $>510 \text{ nm}$ . Additional spectral analysis (excitation at  $370 \text{ nm}$ , emission at  $420\text{--}470 \text{ nm}$ ) indicated a very small fluorescence maximum at  $450 \text{ nm}$  (not shown).

Titration curves were done using other purified GFP mutant proteins to determine whether the pH-dependent spectral properties could be modified by mutagenesis. Fig. 1 C shows that the fluorescence of GFP-Y66H, in which histidine replaces tyrosine in the triamino acid chromophore, was blue-shifted as reported previously (Heim et al., 1994). The spectral intensity decreased with lowered pH, with 50% of maximal intensity at pH  $\sim 6.0$ . In contrast, 50% of maximal intensity was found at pH  $\sim 5$  for GFP-T203I (Fig. 1 D). Absorbance spectra for Y66H and T203I showed single maxima near their corresponding fluorescence excitation maxima, with parallel pH-dependent changes in absorbance and fluorescence (not shown).

The pH titration data for GFP-S65T are summarized in Fig. 2 A. The pH-dependent changes in fluorescence intensity closely paralleled absorbance changes, indicating that it is the GFP-S65T molar absorbance rather than quantum yield (also see fluorescence lifetime data below) that

changes with pH. The titration data were fitted to the following equation:

$$F = A + B/[1 + 10^{n_H(pK_a - pH)}] \quad (1)$$

with parameters  $pK_a$  (pH at 50% maximum) and Hill coefficient  $n_H$  (proportional to slope of fluorescence versus pH at  $pK_a$ ). Parameters A and B are related to signal baseline and gain. Fitted  $pK_a$  and  $n_H$  were 5.91 and 0.92, respectively. A titration of fluorescein (dashed curve) is shown for comparison. The similar shape of the titration curve for GFP-S65T suggests the involvement of a single amino acid residue in its pH-sensitive mechanism. Fluorescence intensity titrations of Y66H and T203I are shown in Fig. 2 B, along with a titration of F64L/S65T, the GFP-S65T variant with humanized codon usage that also contains an F64L mutation. Fluorescence spectra (not shown) and the pH titration data for GFP-S65T and GFP-F64L/S65T were essentially indistinguishable. Fitted  $pK_a$  and  $n_H$  values were 5.98 and 0.97 (GFP-F64L/S65T), 5.98 and 0.65 (GFP-Y66H), and 5.05 and 0.91 (GFP-T203I). The Hill coefficient of under unity for Y66H indicates negative cooperativity, suggesting involvement of more than one residue in its pH-sensitive mechanism. These results indicate that mutagenesis can significantly alter GFP pH dependence.

Additional spectroscopic and kinetic studies were done to investigate the mechanism of the GFP pH sensitivity. Fre-

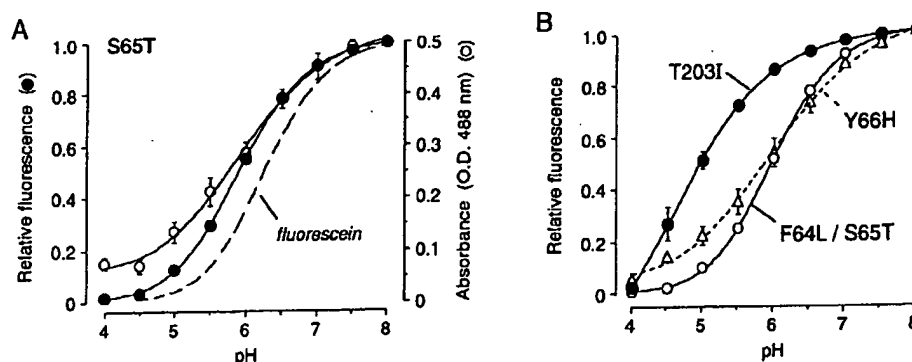


FIGURE 2 pH titrations of GFPs. (A) Fluorescence and absorbance of GFP-S65T as a function of pH. Data were fitted to Eq. 1 with  $pK_a$  and  $n_H$  parameters given in the text. For comparison, fluorescence titration for fluorescein is shown (---). (B) Fluorescence of GFP-F64L/S65T, GFP-Y66H, and GFP-T203I as a function of pH with curve fits as in A.

quency-domain fluorimetry showed that GFP-S65T has a single fluorescence lifetime of  $\sim 2.8$  ns that is independent of pH over a wide range of pH values (Fig. 3 A). This result supports the conclusion that pH affects GFP molar absorbance rather than quantum yield. Circular dichroism spectra for GFP-S65T at pH 7.0 and 5.0 did not differ significantly (Fig. 3 B). Acrylamide did not quench GFP-S65T fluorescence (Stern-Volmer constant  $< 2 \text{ M}^{-1}$ ) at pH 5–7 (not shown). Together these results indicate that major changes in secondary structure content do not occur over a pH range in which fluorescence intensity changes by a factor of  $\sim 15$ .

The reversibility of GFP-S65T fluorescence with pH was studied by titrating pH between 6.5 and different pH values (Fig. 3 C). After correcting for dilution, GFP-S65T fluorescence was found to change nearly reversibly between pH 6.5 and pH values down to  $\sim 5.0$ . Fluorescence was not completely reversed at lower pH ( $< 5$ ), possibly because of a conformational change in the GFP protein and/or decreased solubility. Stopped-flow kinetic measurements were done to distinguish between simple protonation reactions, which would occur in  $< 1$  ms, and protein conformation changes or denaturation, which could occur over millise-

cond and longer times. Nearly all GFP-S65T fluorescence changed in  $< 1$  ms in response to a change in solution pH from 6 to 7 and 7 to 6 (Fig. 3 D). An additional slower process ( $t_{1/2}$  of  $\sim 1$  s) was seen for a change in pH from 5 to 3. Together, these results suggest that for pH  $> 5$ , simple protonation of residue(s) on GFP-S65T produces a decrease in molar absorbance at 480 nm and a consequent decrease in fluorescence.

To evaluate the suitability of GFP as a pH indicator in living cells, fluorescence measurements were carried out in cells expressing GFP-F64L/S65T in cytoplasm and various intracellular compartments. Cells were transfected with cDNAs encoding GFP-F64L/S65T alone (for cytoplasmic staining) or in fusion with appropriate organelle targeting sequences as described in Materials and Methods. The fluorescence micrographs in Fig. 4 show bright and specific staining of the targeted cellular compartments. Subsequent measurements of fluorescence were made by microscopy using a photomultiplier to integrate total fluorescence from groups of 10–20 cells.

Intracellular titrations of GFP-F64L/S65T fluorescence versus pH were done using ionophores to equalize intracel-

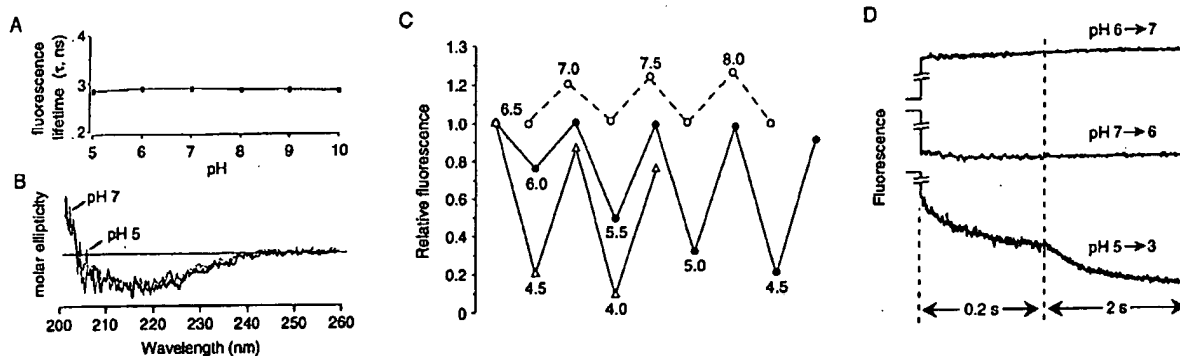


FIGURE 3 pH-dependent properties of GFP-S65T. (A) Fluorescence lifetimes of GFP-S65T ( $\sim 50 \mu\text{g/ml}$  in buffer A) measured by phase-modulation fluorimetry. (B) Circular dichroism spectra of solutions from A. (C) Reversibility of GFP-S65T fluorescence with pH changes. Fluorescence was measured in a stirred cuvette. Where indicated, pH changes were established by addition of microliter aliquots of 1 M HCl or NaOH. (D) Stopped-flow kinetic measurement of time course of GFP-S65T fluorescence in response to indicated changes in pH. All measurements were done at  $23^\circ\text{C}$ .



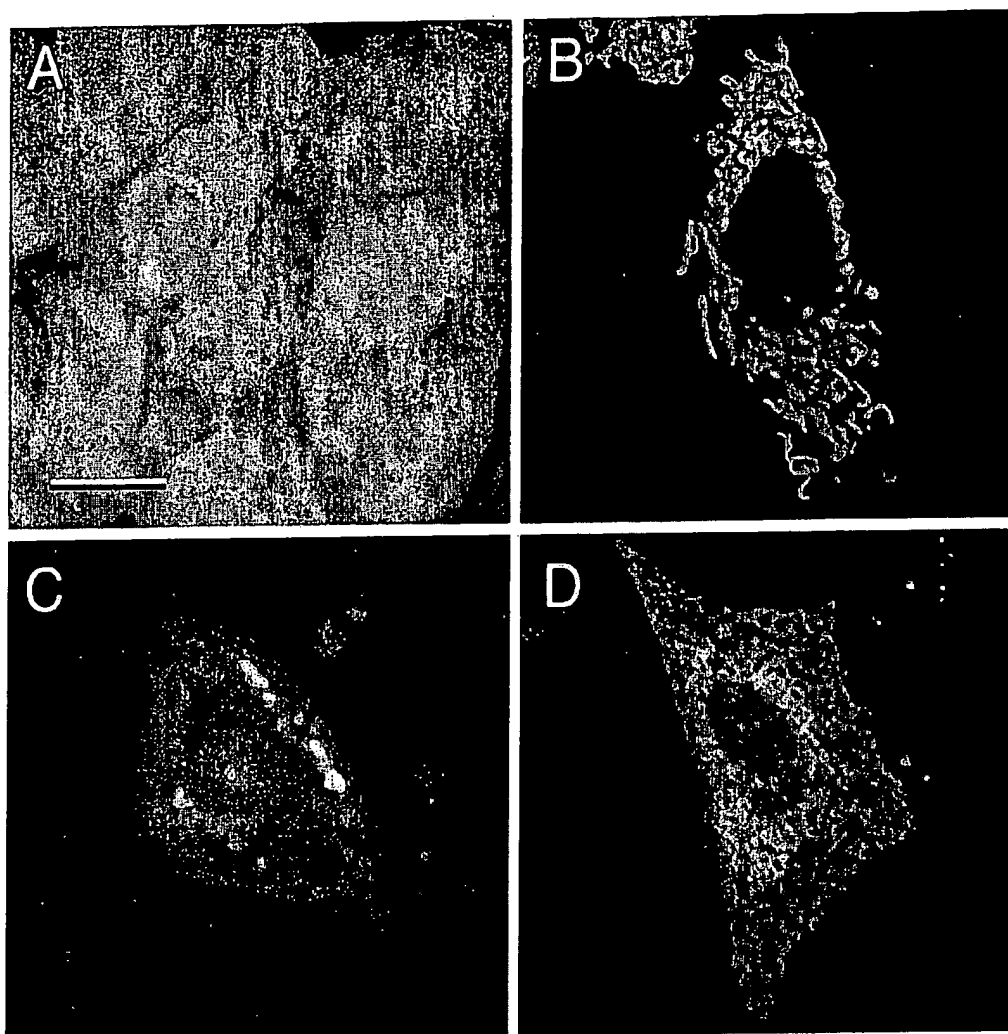


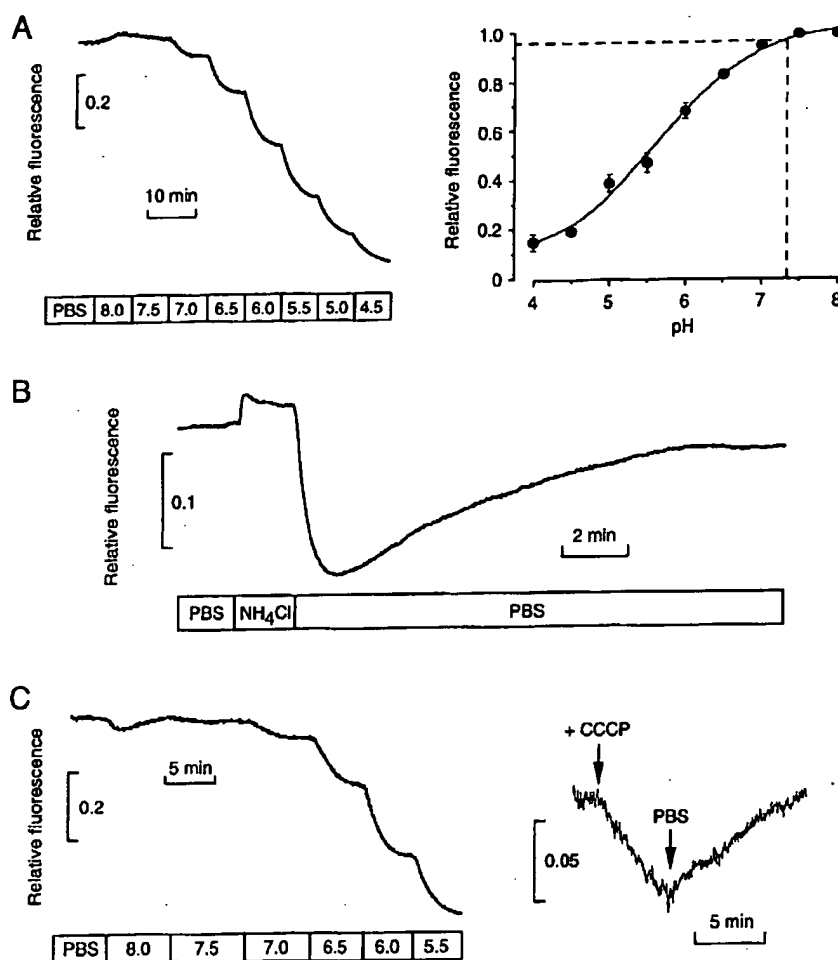
FIGURE 4 Fluorescence micrographs of GFP-F64L/S65T in the cytoplasm of LLC-PK1 cells (A) and the mitochondria (B), Golgi (C), and endoplasmic reticulum (D) of CHO cells. Cells were transfected with GFP-F64L/S65T in fusion with appropriate targeting sequences as described in Materials and Methods.

lular and extracellular pH. Coverglasses containing cells were mounted in a laminar-flow chamber in which perfusion solutions could be exchanged in  $<1$  s. Fig. 5 A (left) shows a representative titration (for GFP-F64L/S65T in cytoplasm) of fluorescence in response to extracellular perfusion with solutions of indicated pH containing ionophores. Large and reversible (not shown) pH-dependent changes in fluorescence were observed. There was little photobleaching and no evidence of photodynamic injury over  $>1$  h using low illumination. Fig. 5 A (right) summarizes averaged pH titration data for four cell preparations. The titration was nearly identical to that for purified GFP-F64L/S65T in saline. A nearly identical titration was obtained using digitonin in place of the ionophores (not shown). Averaged cytoplasmic pH, determined from the fluorescence during perfusion with PBS (not containing ionophores), was  $7.35 \pm 0.03$ .

The response kinetics of cytoplasmic GFP-F64L/S65T fluorescence was measured by addition and removal of  $\text{NH}_4\text{Cl}$ , which promptly alkalinizes (upon addition) and acidifies (upon removal) intracellular compartments because of rapid  $\text{NH}_3$  transport and  $\text{NH}_3/\text{NH}_4^+$  equilibration. Fig. 5 B shows that  $\text{NH}_4\text{Cl}$  addition (under isosmolar conditions) produces a rapid rise in fluorescence due to intracellular alkalinization, followed by a slower decrease in fluorescence resulting from  $\text{NH}_4$  transport and pH regulation. Subsequent replacement of  $\text{NH}_4\text{Cl}$  by  $\text{NaCl}$  produced a prompt intracellular acidification followed by a slower regulatory phase. The greater magnitude of the fluorescence decrease is due to the nonlinear dependence of GFP-F64L/S65T fluorescence on pH (Fig. 5 A). Therefore, intracellular GFP fluorescence responds rapidly to pH changes.

Experiments were done to demonstrate the usefulness of GFP as an indicator of intraorganellar pH changes. Fig. 5 C

FIGURE 5 GFP as an intracellular pH indicator. (A) Titration of fluorescence versus pH in LLC-PK1 cells expressing GFP-F64L/S65T. Cells were initially perfused with PBS and then with buffer B containing ionophores at indicated pH (see Materials and Methods); (left) Representative titration curve; (right) Averaged fluorescence (SE,  $n = 4$ ) with fit to Eq. 1. (B) Time course of cytoplasmic GFP-F64L/S65T fluorescence in response to replacement of 30 mM NaCl by  $\text{NH}_4\text{Cl}$  and subsequent return to NaCl. (C, left) Titration as in A for CHO cells expressing GFP-F64L/S65T in mitochondria; (right) Response of mitochondrial GFP-F64L/S65T fluorescence to addition of the protonophore CCCP (10  $\mu\text{M}$ ) to the PBS perfusate.



(left) shows a titration of fluorescence versus pH for GFP-F64L/S65T in the mitochondrial matrix. The pH-dependent fluorescence was similar to that for GFP-F64L/S65T in cytoplasm. Mitochondrial pH was relatively high ( $>7.5$ ) but could not be determined accurately because of the much lower  $\text{pK}_a$  of GFP-F64L/S65T. A transient decrease in signal was observed upon switching from PBS to the pH 8.0 calibration solution containing ionophores. This decrease is probably due to an increase in proton conductance of the mitochondrial membranes, resulting in proton influx driven by the strong interior negative mitochondrial membrane potential. Fig. 5 C (right) shows that addition of the protonophore CCCP (without other components of the calibration solution) produced reversible acidification of the mitochondrial lumen.

Fig. 6 A shows a pH titration for CHO cells expressing GFP-F64L/S65T in the Golgi compartment. The precise localization of GFP-F64L/S65T (cis- versus medial versus trans-Golgi) was not determined in this study. Addition of the vacuolar proton pump inhibitor bafilomycin A1 produced a slow alkalinization, consistent with results obtained when the trans-Golgi lumen was labeled by liposome fusion with a fluorescein-based pH indicator (Seksek et al., 1995)

and by retrograde transport of a fluorescein-labeled verotoxin receptor (Kim et al., 1996). Subsequent perfusion with calibration solutions showed a reversible change in fluorescence similar to results in cytoplasm and mitochondria. Fig. 6 B shows the kinetics of Golgi GFP-F64L/S65T fluorescence in response to addition and removal of the weak acid acetate and the weak base  $\text{NH}_4\text{Cl}$ . Acetate addition produced a prompt acidification (because of rapid acetic acid influx and dissociation) and slower alkalinization. Subsequent acetate removal gave a prompt alkalinization.  $\text{NH}_4\text{Cl}$  addition and removal produced alkalinization and acidification, respectively, as seen for GFP-F64L/S65T in cytoplasm in Fig. 5 B.

## DISCUSSION

Unique advantages of GFP as a targeted pH indicator are the ability to measure pH at specific intracellular sites with little background signal and no indicator leakage, and without the toxicities associated with chemical indicators and invasive loading procedures. An ideal fluorescent pH sensor should have high pH sensitivity and specificity, rapid signal re-

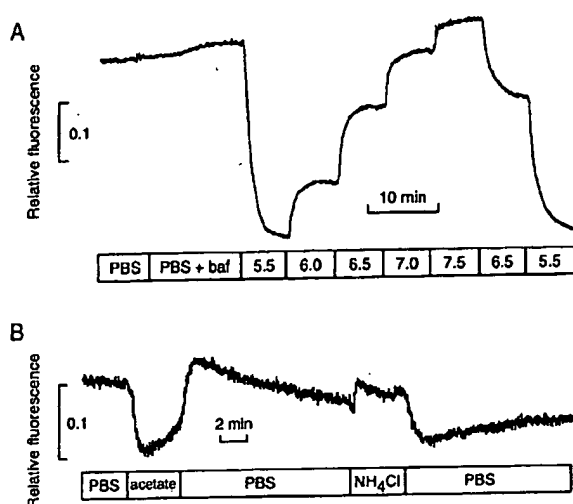


FIGURE 6 Measurement of Golgi changes pH by GFP-F64L/S65T fluorescence. (A) Titration of fluorescence versus pH in CHO cells expressing GFP-F64L/S65T in Golgi. Cells were initially perfused with PBS and then with PBS containing 10 nM bafilomycin A1, followed by calibration solutions (buffer B) containing ionophores at indicated pH. (B) Kinetics of GFP-F64L/S65T fluorescence in response to replacement of 30 mM NaCl in PBS with 30 mM sodium acetate and  $\text{NH}_4\text{Cl}$ .

sponse to pH changes, and good optical properties. The results here indicate that GFP fulfills many of these requirements. GFP fluorescence changes by  $\sim 50\%$  for a 1 pH unit change around its apparent  $\text{pK}_a$ , similar to the sensitivity of fluorescein-based and other chemical pH indicators. Although the  $\text{pK}_a$  values of the GFP mutants analyzed here were in the range 5–6, it is likely that the pH sensitivity can be modified by further mutagenesis. Because of the protected environment of the GFP chromophore, its fluorescence should be highly selective for pH; GFP fluorescence and other optical properties are insensitive to quenchers and other chemical agents that strongly influence the optical properties of many chemical chromophores (Swaminathan et al., 1997). GFP fluorescence responds very rapidly to pH changes. The available GFP mutants are brightly fluorescent and have excitation and emission maxima at visible wavelengths, where autofluorescence and photodynamic injury are minimal. Although the pH insensitivity of spectral shape precludes ratio imaging of existing GFPs, it should be possible to couple GFPs with different emission wavelengths and  $\text{pK}_a$  values to generate a pH indicator suitable for ratio imaging. For cell studies, GFP is an ideal indicator because it is nontoxic, chemically inert, and targetable to selected intracellular locations without leakage or migration.

The pH sensitivity of GFP has implications for its use as a sensor for parameters other than pH. Fluorescence energy transfer between GFP mutants has recently been used in the design of protein sensors to measure protease activity (Heim and Tsien, 1996) and calcium concentration (Romoser et al., 1997; Miyawaki et al., 1997). Because of the pH sensitivity of GFP fluorescence and absorbance, the Forster distance and thus the sensor response will also depend on pH.

Apparent calcium concentrations would thus be different in acidic cellular compartments. GFP mutants with  $\text{pK}_a$  values much lower than the pH of the sensor environment are needed to eliminate pH effects.

Several experiments suggested that the pH sensitivity of GFP fluorescence at pH  $> 5$  results from protonation-deprotonation of residues at or near the chromophore. The response of GFP fluorescence to pH changes occurred in  $< 1$  ms and was reversible. GFP fluorescence lifetime, fluorescence spectral shape, CD spectra, and fluorescence quenching by acrylamide were also insensitive to pH. At pH  $< 5$ , GFP fluorescence responded relatively slowly to pH changes, and the response was not completely reversible. It is thus likely that the GFP unfolds at very low pH, limiting its usefulness as a pH sensor in highly acidic environments.

Crystallographic and spectroscopic data suggest that the protonation state of the phenolic group of the chromophore is responsible for the GFP pH sensitivity. The chromophore in denatured, wild-type GFP has pH-dependent spectral characteristics due to the ionization of the tyrosine 66 phenolic group (Ward et al., 1982). The phenolate form of the chromophore has an absorption maximum at 448 nm compared with that of 384 nm for the uncharged phenol. The  $\text{pK}_a$  for this transition is 8.1. Based on the excited state dynamics of GFP, it was proposed that wild-type GFP can exist in one of two ground states, A and B, which differ in protonation state of the chromophore (Chatteraj et al., 1996). An excited-state proton transfer reaction rapidly converts state A to intermediate state I, which is slowly converted to state B. State A absorbs at 404 nm and emits at 420–470 nm, state I emits at 500 nm, and state B absorbs at 471 nm and emits at 482 nm. Crystallographic studies confirmed the existence of two ground state conformations each with distinct spectral characteristics (Brjcek et al., 1997; Palm et al., 1997). In GFP mutants with excitation maxima at  $\sim 395$  nm, the phenol in tyrosine 66 is uncharged (corresponding to state A), whereas it is in the charged phenolate form in mutants with excitation maxima at 473 nm (corresponding to state B).

Our findings suggest that pH shifts the equilibrium between the GFP A and B ground states. At high pH, the phenolate form of tyrosine is favored so that the B state is populated and excitation and emission occur near 471 nm and 500 nm, respectively (Fig. 1 A). At low pH, the phenol form is favored so that state A is populated and absorbance shifts to 390 nm (Fig. 1 B). The absence of fluorescence emission at 500 nm by excitation at 390 nm is due either to quenching of the I state or to the inability of the I state to convert to the B state. The lower  $\text{pK}_a$  of 6 for the phenol-phenolate transition in folded GFP compared with denatured GFP is probably due to stabilization of the phenolate form by the network of hydrogen bonds in the folded protein.

The nonunity Hill coefficient for the fluorescence response in Y66H suggests that more than one titratable residue influences the relative stability of the two states. The pH sensitivity for this mutant probably occurs by a mech-

anism in which a water molecule replaces the phenolic oxygen of tyrosine 66 (Palm et al., 1997). Additional work is needed to identify other mutations in GFP that alter  $pK_a$  by shifting the equilibrium between the charged and uncharged forms.

The analysis of GFP fluorescence in cell cytoplasm indicated that the dependence of GFP fluorescence on pH was similar to that of the purified protein in saline. Ammonium chloride pulse studies indicated that the GFP signal response is very fast and demonstrated the expected recovery phase involving pH regulatory mechanisms (Roos and Boron, 1981). Continuous measurements of cytoplasmic GFP fluorescence could be made with excellent signal-to-noise ratio and little photobleaching. Fluorescence versus pH calibrations for GFP in various organellar compartments confirmed that the sensitivity of GFP fluorescence to pH did not depend on its location in cells. The experiments in Figs. 5 and 6 demonstrated the ability to follow intracellular pH changes in response to various maneuvers.

The use of GFP as an intracellular pH indicator should permit many types of measurements that cannot easily be accomplished using existing chemical pH indicators. The targeting of GFP to the lumen of organelles allows direct measurements of intraorganellar pH and analysis of pH regulatory mechanisms. There is evidence that trans-Golgi pH is regulated by second messengers including cAMP (Seksek et al., 1995), but the mechanistic basis of this regulation is unknown. Little information is available on pH changes in endoplasmic reticulum, medial and cis-Golgi, mitochondria, and other intracellular compartments. There is provocative evidence that intranuclear pH might be higher than cytoplasmic pH (Seksek and Bolard, 1996); however, the measurements involved indirect chemical indicators with uncertainties in indicator calibration and optical properties. It is thought that chloride channels on organellar membranes are needed to shunt charge to permit the generation of an acidic luminal pH. Intracellular ion transporting mechanisms such as chloride conductance can be measured by coupling ion movement to electrogenic proton transport (by use of ionophores, Biwersi et al., 1994). Finally, selective GFP targeting in transgenic mice should permit measurements of cellular pH in vivo.

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